

# B.M.S. College of Engineering, Bengaluru-560019

Autonomous Institute Affiliated to VTU

## April 2023 Semester End Main Examinations

**Programme: B.E.**

**Branch: ME/CH/AS/IEM**

**Course Code: 22CHIBSCME**

**Course: Applied chemistry for Mechanical Engineering stream**

**Semester: I**

**Duration: 3 hrs.**

**Max Marks: 100**

**Date: 08.04.2023**

**Instructions:** 1. Answer any FIVE full questions, choosing one full question from each unit.  
2. Missing data, if any, may be suitably assumed.

### UNIT - I

- 1 a) With a neat diagram, explain the construction and working of Calomel electrode. **05**
- b) Two copper rods are placed in 0.1M and 1.0 M CuSO<sub>4</sub> solutions separately to form a cell. Write the representation of the cell. Evaluate the EMF of the cell at 25°C. **04**
- c) Explain the corrosion of iron based on electrochemical theory. **06**
- d) Describe the electroless plating of copper with relevant reactions. **05**

### OR

- 2 a) Describe the experimental details and give the mathematical derivation for determination of the pH of a solution using glass electrode. **06**
- b) Define CPR. Calculate CPR in mpy for a thick steel sheet of area 50 inch<sup>2</sup> which experiences a weight loss of 380 g after 365 day. (Density of steel = 7.9 g/cm<sup>3</sup>). **04**
- c) What is cathodic protection? With neat diagram, describe the corrosion control by sacrificial anode method. **06**
- d) List the differences between electro plating and electroless plating. **04**

### UNIT - II

- 3 a) Explain the following battery characteristics: **04**  
(i) Capacity (ii) Energy Efficiency
- b) With a suitable reaction, explain the synthesis of bio-diesel. **04**
- c) Define calorific value. Explain the determination of calorific value of solid fuel by Bomb Calorimeter. **06**
- d) Discuss the Fluidized bed catalytic cracking of petroleum. **06**

**Important Note:** Completing your answers, compulsorily draw diagonal cross lines on the remaining blank pages. Revealing of identification, appeal to evaluator will be treated as malpractice.

### UNIT - III

- 4 a) How is PMMA synthesized? Mention its applications. **04**  
b) Define the following terms: i) Elastomers ii) Polymer composites **06**  
iii) Biodegradable polymers and iv) PDI  
c) Elaborate on the structural property relationship of polymers with reference to **06**  
i) Strength and ii) Chemical resistance.  
d) Give the synthesis and applications of butyl rubber. **04**

### OR

- 5 a) Illustrate any three factors which affect the glass transition temperature. **06**  
b) Evaluate the number average molecular weight and weight average molecular **04**  
weight of a polymer consisting of 10% by weight of macromolecules of  
molecular mass 1000 and 90% by weight of macromolecules of molecular  
mass 10,000.  
c) Explain the synthesis i) Kevlar ii) Nitrile rubber Mention their applications **06**  
d) Distinguish between thermoplastic and thermosetting plastic. **04**

### UNIT - IV

- 6 a) Outline the classification of lubricants? Discuss the applications of the **06**  
lubricants.  
b) What are shape memory alloys? Mention their industrial importance. **04**  
c) Explain the synthesis of nanomaterials by sol-gel method and mention its **06**  
advantages.  
d) Explain the top down and bottom up process for the synthesis of **04**  
nanomaterials.

### UNIT - V

- 7 a) Explain the determination of total hardness of water by complexometric **06**  
method.  
b) Analyze the variation in conductance for the titration of mixture of strong acid **04**  
and weak acid against strong base.  
c) Discuss the phase diagram of water system with the neat diagram. **06**  
d) Describe the various components of colorimeter. **04**

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