

B.M.S. College of Engineering, Bengaluru-560019

Autonomous Institute Affiliated to VTU

August 2024 Semester End Main Examinations

Programme: B.E.

Branch: Chemical Engineering

Course Code: 23CH4ESANI

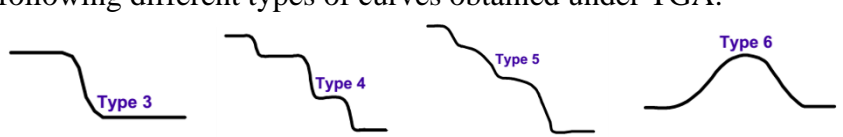
Course: Analytical Instruments

Semester: IV

Duration: 3 hrs.

Max Marks: 100

Instructions: 1. Answer any FIVE full questions, choosing one full question from each unit.
2. Missing data, if any, may be suitably assumed.

Important Note: Completing your answers, compulsorily draw diagonal cross lines on the remaining blank pages. Revealing of identification, appeal to evaluator will be treated as malpractice.			UNIT - I	CO	PO	Marks
	1	a)	Differentiate between qualitative and quantitative analysis. A farmer brought several varieties of potatoes and wants to test for the presence of iodine qualitatively. Describe how he can identify iodine using classical methods.	CO 1	PO 1	10
		b)	Explain the external standard calibration method in analytical chemistry. Discuss its advantages and limitations	CO 2	PO 2	10
			UNIT - II			
	2	a)	Explain the origins of light absorptions in UV-vis and IR Spectroscopy.	CO 1	PO 1	10
		b)	The absorbance of a Cu sulphate solution containing 0.5 mg Cu/ml was reported as 0.35 at 440 nm. What will be the absorbance if the solution is diluted to twice its original volume? Assume that a 1 cm cuvette was used.	CO 1	PO 1	05
		c)	Discuss the deviations in Beer's Law.	CO 1	PO 1	05
			OR			
	3	a)	Draw a neat labelled instrumentation diagram for UV-vis and FTIR spectrometry.	CO 1	PO 1	10
		b)	List out the different types of detectors used in IR spectroscopy and explain any one in detail with a neat sketch.	CO 2	PO 2	10
			UNIT - III			
	4	a)	Give a brief about the information that can be obtained about a sample using Thermogravimetric analysis (TGA). Interpret the following different types of curves obtained under TGA: 	CO 1	PO 1	10

	b)	Write the working principles of Bomb calorimeter and Power compensated differential calorimeter along with their rough sketches.	CO 1	PO 1	10
		UNIT - IV			
5	a)	Explain the principle on which GC works Draw the neat labelled instrumentation diagram of Gas Chromatography (GC) equipment.	CO 1	PO 1	10
	b)	Describe the working of electron capture detector (ECD) used in gas chromatography with a neat sketch.	CO 1	PO 1	10
		OR			
6	a)	Discuss the working of split and split-less injection methods used in Gas Chromatography (GC).	CO 2	PO 2	10
	b)	Explain the working of Flame ionization detector (FID) used in gas chromatography with a neat sketch.	CO 2	PO 2	10
		UNIT - V			
7	a)	Discuss the criteria for selecting mobile phase in High pressure liquid chromatography (HPLC). Draw a neat labelled diagram of HPLC system.	CO 3	PO 2	10
	b)	Enlist the different types of columns used in High pressure liquid chromatography (HPLC). Explain the any two columns in detail.	CO3	PO 2	10
