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# B.M.S. College of Engineering, Bengaluru-560019

Autonomous Institute Affiliated to VTU

## August 2024 Semester End Main Examinations

**Programme: B.E.**

**Semester: IV**

**Branch: Electronics and Instrumentation Engineering**

**Duration: 3 hrs.**

**Course Code: 23EI4PCTNI**

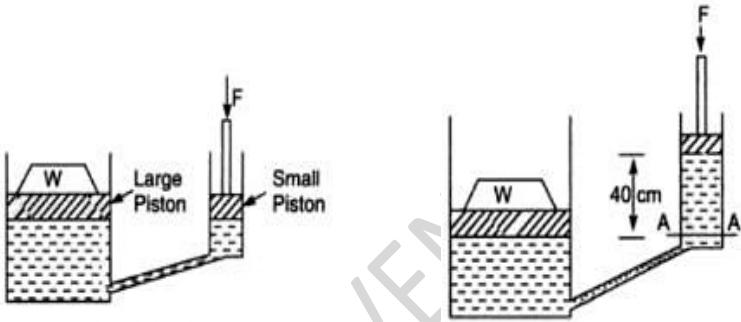
**Max Marks: 100**

**Course: Transducers and Instrumentation**

**Instructions:** 1. Answer any FIVE full questions, choosing one full question from each unit.  
2. Missing data, if any, may be suitably assumed.

<b>MODULE - I</b>			<b>CO</b>	<b>PO</b>	<b>Marks</b>
1	a)	Differentiate between transducer and sensor with example.	<i>CO1</i>	<i>PO1</i>	<b>05</b>
	b)	Explain Different IO Configurations	<i>CO1</i>	<i>PO1</i>	<b>05</b>
	c)	Illustrate with block diagram, the functional elements of an instrument system.	<i>CO1</i>	<i>PO1</i>	<b>10</b>
<b>MODULE- II</b>					
2	a)	Illustrate the working principles of an Orifice Plate Flow Meter by applying its design features to real-world scenarios.	<i>CO2</i>	<i>PO2</i>	<b>10</b>
	b)	An oil of viscosity $0.5 \text{ Ns/m}^2$ and relative density 0.6 is flowing through a circular pipe of diameter 60 cm and of length 500 m. The average velocity of the oil is 2 m/s. Determine the Reynolds number and state whether the flow is laminar or turbulent	<i>CO2</i>	<i>PO3</i>	<b>05</b>
	c)	Through a refinery, fuel ethanol is flowing in a pipe at a velocity of 1 m/s and a pressure of 101300 Pa. The refinery needs the ethanol to be at a pressure of 2 atm (202600 Pa) on a lower level. How far must the pipe drop in height in order to achieve this pressure? Assume the velocity does not change. The density of ethanol is $789 \text{ kg/m}^3$ and gravity $g$ is $9.8 \text{ m/s}^2$ .	<i>CO2</i>	<i>PO3</i>	<b>05</b>
<b>OR</b>					
3	a)	Explain with the help of a neat diagram the working of an Rotameter and describe its advantages and disadvantages	<i>CO2</i>	<i>PO2</i>	<b>10</b>
	b)	With a neat diagram, Explain the working of an Electromagnetic flow meter	<i>CO2</i>	<i>PO2</i>	<b>10</b>
<b>MODULE- III</b>					
4	a)	PT-100 is a Platinum RTD whose resistance at $0^\circ\text{C}$ is 100ohm. If $3.91 \times 10^{-3}/^\circ\text{C}$ is the resistance temperature co-efficient of Platinum, then find its resistance at $100^\circ\text{C}$ .	<i>CO3</i>	<i>PO3</i>	<b>05</b>

**Important Note:** Competing your answers, compulsorily draw diagonal cross lines on the remaining blank pages. Revealing of identification, appeal to evaluator will be treated as malpractice.

		b)	A thermocouple has a linear sensitivity of 30uv/C, calibrated at a cold junction temperature of 0°C. It is used to measure an unknown temperature with the cold junction temperature of 30°C. Find the actual hot junction temperature if the emf generated is 3.0 mv.	CO3	PO3	<b>05</b>
		c)	Mention the different types of Thermocouple available and explain its characteristics	CO3	PO2	<b>10</b>
<b>MODULE - IV</b>						
5	a)	Explain in detail, with a neat sketch the working of McLeod Vacuum Gauge	CO4	PO2	<b>10</b>	
	b)	The diameters of a small piston and a large piston of a hydraulic jack are 3 cm and 10 cm respectively. A force of 80 N is applied on the small piston. Find the load lifted by the large piston when: (a) the pistons are at the same level. (b) small piston is 40 cm above the large piston. The density of the liquid in the jack is given as 1000 kg/m <sup>3</sup>	CO4	PO3	<b>10</b>	
		 <p>Figure 5(b)</p>				
		<b>OR</b>				
6	a)	With a neat sketch, Explain the working principle of a dead weight tester	CO4	PO2	<b>10</b>	
	b)	Write a note on Elastic type Pressure Gauges	CO4	PO2	<b>10</b>	
<b>MODULE - V</b>						
7	a)	Detail the step-by-step process of designing a cloud-based home automation system, incorporating a framework that outlines each component's role within the system. What will be the conceptual equation for the application?	CO4	PO3	<b>12</b>	
	b)	What is a IoT? Break down the individual technologies behind the Internet of Things, examining their roles, interdependencies, and how they collectively contribute to the seamless functioning of IoT applications.	CO4	PO3	<b>08</b>	

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