

B.M.S. College of Engineering, Bengaluru-560019

Autonomous Institute Affiliated to VTU

September / October 2024 Supplementary Examinations

Programme: B.E.

Branch: Mechanical Engineering

Course Code: 20ME6DCFHT

Course: Fundamentals of Heat Transfer

Semester: VI

Duration: 3 hrs.

Max Marks: 100

- Instructions:**
1. Answer any FIVE full questions, choosing one full question from each unit.
 2. Missing data, if any, may be suitably assumed.
 3. Use of Heat Transfer Data Hand Book is permitted.

Important Note: Completing your answers, compulsorily draw diagonal cross lines on the remaining blank pages. Revealing of identification, appeal to evaluator will be treated as malpractice.			UNIT - I	CO	PO	Marks
	1	a)	State the assumptions and derive general 3-dimensional heat conduction equation in Cartesian co-ordinates.	CO2	PO1	10
		b)	A metallic steam pipe ($k=45 \text{ W/m}^0\text{C}$) 5 cm ID and 6 cm OD is lagged with a 2.75 cm radial thickness of high temperature insulation having thermal conductivity of $1.1 \text{ W/m}^0\text{C}$. The surface heat transfer coefficient for inside and outside are $4650 \text{ W/m}^2\text{K}$ and $11.5 \text{ W/m}^2\text{K}$ respectively. If steam temperature is at 200^0C and the ambient temperature is 25^0C . Determine heat loss per meter length of pipe and temperature at the interfaces.	CO2	PO2	10
			OR			
	2	a)	Derive an expression for temperature distribution for a infinite long fin of uniform cross section.	CO2	PO1	08
		b)	Write the significance of Biot number and Fourier number in transient heat conduction.	CO3	PO1	04
		c)	A stainless steel rod of outer dia. 1 cm originally at a temperature of 320^0C is suddenly immersed in a liquid at 120^0C for which the convective heat transfer coefficient is $100 \text{ W/m}^2\text{K}$. Determine the time required for the rod to reach a temperature of 200^0C .	CO3	PO1	08
			UNIT - II			
	3	a)	Explain the formation of velocity and thermal boundary layers for flow over a flat plate. Draw the profile curves.	CO4	PO1	10
		b)	Atmospheric air at 25^0C flows over both the surfaces of a flat plate 1 m long with a velocity of 5 m/s. The plate is maintained at a uniform temperature of 75^0C . Determine, i) the velocity boundary layer thickness, the shear stress and heat flux at the trailing edge of the plate, ii) the drag force on the plate and the total heat transfer from the plate to air.	CO4	PO4	10

		OR			
4	a)	Explain the physical significance of Reynolds number, Grashoff number, Prandtl number, Nusselt number and Peclet number.	CO4	PO1	10
	b)	Lubricating oil at a temperature of 60°C enters a 1cm diameter tube with a velocity of 2.5 m/s. The tube surface is maintained at 30°C. Calculate the length of tube required to cool the oil to 45°C. Assume the oil has the following average properties, $\rho=865 \text{ kg/m}^3$, $\mu=7.75 \times 10^{-3} \text{ kg/ms}$, $C=1.6 \text{ kJ/kgK}$, $k=0.12 \text{ W/mK}$. Use $Nu=0.023Re^{0.8}Pr^{0.4}$.	CO4	PO2	10
		UNIT - III			
5	a)	With the application of dimensional analysis for a free convection obtain the correlation, $Nu=C Gr^a Pr^b$	CO4	PO1	12
	b)	A hot plate 1 m x 0.5 m at 130°C is kept vertically in still air at 20°C. Find, i) heat transfer coefficient and ii) heat lost to surroundings.	CO4	PO2	08
		UNIT - IV			
6	a)	State and explain, Planck's law, Wien's displacement law, and Kirchhoff's law.	CO6	PO1	09
	b)	Explain the concept of black body.	CO6	PO1	04
	c)	The temperature of black surface of 0.2 m ² area is 540°C. Calculate, i) Total rate of energy emission, ii) Intensity of normal radiation, and iii) Wavelength of maximum monochromatic emission power.	CO6	PO1	07
		UNIT - V			
7	a)	Derive the expression for LMTD for a counter flow heat exchanger. List out the assumptions made.	CO5	PO1	10
	b)	Oil at 100°C ($C_p=3.6 \text{ kJ/kgK}$) flows at a rate of 30000 kg/hr and enters a parallel flow heat exchanger. Cooling water ($C_p= 4.2 \text{ kJ/kgK}$) enters a heat exchanger at 10°C at rate of 50000 kg/hr, the area of HE is 10 m ² , $U=1000 \text{ W/m}^2\text{K}$. Calculate the outlet temperature of oil and water.	CO5	PO2	10
